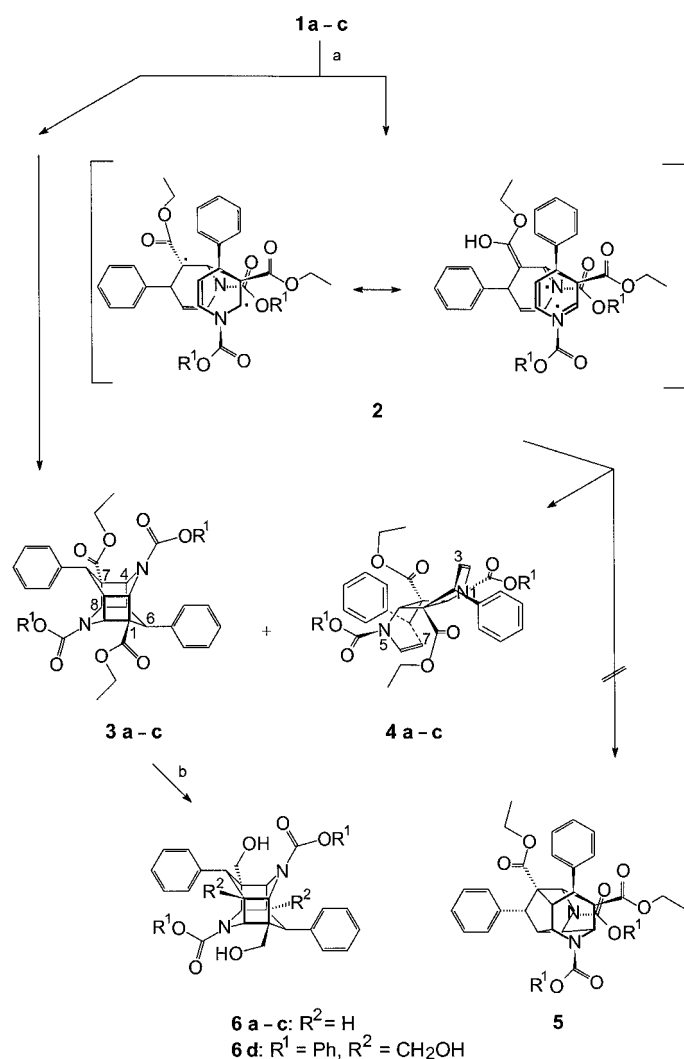






Scheme 2. Synthesis of the target structures **6a–c** starting from the monomeric 4-aryl-1,4-dihydropyridines. a)  $h\nu$ ,  $\lambda > 270$  nm, 4 weeks, 27 °C, THF, **3a** (44 %), **4a** (33 %), **3b** (46 %), **4b** (32 %), **3c** (41 %), **4c** (38 %); b)  $\text{LiAlH}_4$  (1 equiv), 2 h, –8 °C, THF, **6a** (78 %), **6b** (75 %), **6c** (65 %).

kishomocubane **5** was formed.<sup>[7]</sup> We are currently investigating the causes of the unexpected outcome of this reaction. In the case of the *anti* dimers we propose a stabilizing conjugated effect of the *N*-oxycarbonyl group on a hypothetical intermediate biradical **2**, which then undergoes bond rotation to form the *anti* dimer. Subsequent selective reduction of the ester groups at the substituted cyclobutane ring of the dimers is accomplished with lithium aluminum hydride at low temperature and provides the target compounds **6a–c**.

The *in vitro* assay to determine the inhibition of the activity of the molecular pump made use of fluorescence measurements of untreated cells and cells that had been treated with P-GP modulators. The ratio of the uptake of a P-GP transporter substrate that shows specific binding and fluorescence, for example, rhodamine 123, by tumor cells and cells that have become resistant through MDR gene transfer was calculated as the ratio of P-GP inhibitory activity *R* (Table 1).<sup>[8]</sup> This direct comparison of the fluorescence data ensures that the observed activities observed for the specific P-GP inhibitors is based on the MDR modulators.

Table 1. Effects of the hydroxymethyl-3,9-diazatetraasteranes **6a–d** on the MDR of tumor cells (mouse T-lymphoma).<sup>[a]</sup>

| Compound  | Activity ratio <i>R</i> <sup>[b,c]</sup> |                      |
|-----------|------------------------------------------|----------------------|
|           | 3 $\mu\text{M}$                          | 30 $\mu\text{M}$     |
| <b>6a</b> | 1.11                                     | 18.62                |
| <b>6b</b> | 28.10                                    | 30.87 <sup>[d]</sup> |
| <b>6c</b> | 6.81                                     | 7.42 <sup>[d]</sup>  |
| <b>6d</b> | 0.97                                     | 5.40                 |

[a] VRP control with 11  $\mu\text{M}$ : 7.27. According to ref. [8] 11  $\mu\text{M}$  is the relevant maximum concentration for MDR modulation. The compounds are active (A) when the ratio is  $R > 1.1$ , and very active (AA) for *R* values above 10.<sup>[8]</sup> [b]  $R = \frac{(\text{MDR, treated}/\text{MDR, untreated control})}{(\text{parent cell line, treated}/\text{parent cell line, untreated control})}$ . [c] Average value from two determinations. [d] Saturation concentration for P-GP inhibition.

In the case of the hydroxymethyl cage compounds **6a–c**, the *N*-methoxycarbonyl-substituted compound **6a** showed an activity similar to that of VRP; a linear relationship between the concentration and the inhibitory activity was determined. Structural characteristics required for sufficient activity of MDR modulators include the presence of two lipophilic moieties and a basic nitrogen atom, which was replaced for the first time by Ecker et al. by an ester group that functioned as a hydrogen bonding acceptor.<sup>[9,10]</sup> In our compounds the nitrogen atom, a hydrophilic structural element, is replaced by the hydroxymethyl group. The theory that sufficient lipophilicity is requirement for good MDR modulators was supported by the fact that compounds **6b** and **6c**, which contain more lipophilic groups at the *N*-substituents, showed increased activity at low concentrations.<sup>[11]</sup> It is interesting that an increase in the total number of potential hydrogen bond donor and acceptor functionalities from four (two hydroxymethyl and two carbamide ester groups) in **6b** to six (four hydroxymethyl and two carbamide ester groups) in **6d**<sup>[12]</sup> is accompanied by complete loss of activity at the same low concentration (3  $\mu\text{M}$ ).

Until now the presence of a large number of moieties that can participate in hydrogen bonding, for interaction with the amide groups that are in the proposed bonding site, was considered to be important for an optimal bonding affinity to the P-GP.<sup>[11,13,14]</sup> Initial studies on a series of 3,9-dialkyl-3,9-diazatetraasteranes<sup>[15]</sup> provide inhibitory activity ratios of 30.19 at 3  $\mu\text{M}$  and 24.83 at 30  $\mu\text{M}$  for a 3,9-dibenzyl derivative with four hydroxymethyl groups. This corresponds to the activity of compound **6b**, which contains two hydroxymethyl and two carbamide ester groups and therefore also presents a total of four hydrogen bonding sites. A 3,9-dimethyl-3,9-diazatetraasterane with four hydroxymethyl groups provides *R* values of 0.88 (3  $\mu\text{M}$ ) and 4.7 (30  $\mu\text{M}$ ) and thus does not display any noteworthy P-GP inhibitory activity until a concentration of 30  $\mu\text{M}$ . This underlines the importance of a bulky lipophilic *N*-substituent for inhibitory activity, as shown by the comparison of the 3,9-dimethoxycarbonyl derivative **6a** with the 3,9-diphenoxycarbonyl derivative **6b**. A decrease in the number of hydroxymethyl groups from four to two leads to a decrease in the inhibitory activity ratio *R* to 16.42 (3  $\mu\text{M}$ ) and 10.69 for the 3,9-dibenzyl derivative (30  $\mu\text{M}$ ) and to 0.67 (3  $\mu\text{M}$ ) and 2.7 (30  $\mu\text{M}$ ) for the 3,9-dimethyl derivative. In the case of four hydroxymethyl groups, similar results are obtained as for the combination of two hydroxymethyl and

two carbamide ester groups. In contrast, the presence of four hydroxymethyl and two carbamide ester groups, and therefore a total of six hydrogen bond donor and acceptor functionalities, is considered to be less favorable. Our results suggest a limit on the number of hydrogen bond donor and acceptor groups, at least for this first series of symmetric modulators, and therefore provide an important contribution to the discussion on the role of hydrogen bonding with respect to activity. A comparison of the concentrations required for sufficient inhibitory activity ( $< 3 \mu\text{M}$ ) and the  $\text{CC}_{50}$  values for the cytotoxicity of **6b** and **6c** ( $> 93$  and  $125 \mu\text{M}$ , respectively), obtained from MTT assays<sup>[16]</sup> for the determination of lactate dehydrogenase (LDH) activity in metabolically active normal cells (MT-4), makes clear that the novel MDR modulators evoke convincing inhibitory activities at concentrations below those causing cytotoxic effects.

In conclusion, hydroxymethyl-substituted 3,9-diazatetraateranes represent an independent class of novel MDR modulators that show activity at concentrations below cytotoxic levels. A highly interesting characteristic of these compounds is the rigidly symmetric form, which, owing to the extraordinary effect on the target structure, could be a reflection of the composition of the molecular pump or the potential bonding region, whose structure is as yet unknown. The goal of further work is to examine this "symmetry hypothesis" through the synthesis and investigation of unsymmetrical compounds. In order to estimate the clinical potential experiments are planned on additional resistant cell lines, whose MDR is based on the expression of molecular pumps other than the P-GP.

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## Fluorescence Spectroscopic Quantification of the Release of Cyclic Nucleotides from Photocleavable [Bis(carboxymethoxy)coumarin-4-yl]methyl Esters inside Cells\*\*

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Caged compounds are an elegant means to produce rapid jumps in the concentration of chemical messenger molecules inside cells.<sup>[1]</sup> Caged compounds are photolabile inactive derivatives of biologically active molecules. The biologically active substance is rapidly released by a photochemical reaction. Caged compounds allow the elucidation of complex intracellular processes and their resolution in time and space.

For many applications it would be useful to determine quantitatively the degree of photolysis of the caged compound inside cells. In a few cases this has been possible by a rather complicated calibration of the cellular reaction. Here we show that the determination of the amount of released cyclic nucleotides is feasible by fluorescence measurements, using the axial and equatorial diastereomers of [6,7-bis(carboxymethoxy)coumarin-4-yl]methyl (BCMCM) esters of cyclic adenosine-3',5'-monophosphate (cAMP) **1** and cyclic guanosine-3',5'-monophosphate (cGMP) **2**,<sup>[2]</sup> which are excellent phototriggers for cyclic nucleotides.

Upon irradiation with UV light in aqueous buffer solution and inside cells, **1** and **2** rapidly produce (within 2–5 ns) cAMP

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